Another problem at the time had to do with light...

# The Electromagnetic Spectrum 

Visible Light


To be completed in class!
(leave 2-3 lines for labeling below)

$$
c=\lambda v
$$

c = speed of light (all light travels at this speed)

$$
\mathrm{c}=3.0 \times 10^{5} \mathrm{~km} / \mathrm{s}(186,000 \mathrm{mi} / \mathrm{s})
$$

$\lambda=$ wavelength (distance from peak to peak)
$v=$ frequency (number of waves passing a point in a certain time)
wavelength praw this diagram!


## Two Problems

1. Why don't negative electrons crash into the positive nucleus?
2. What causes elements to emit a characteristic spectrum of colored lines?

A new model solved BOTH problems!

