

# Math Essentials

Draw this diagram!

# Precision vs. Accuracy

Accurate  
Precise



Not Accurate  
Precise



Accurate  
Not Precise



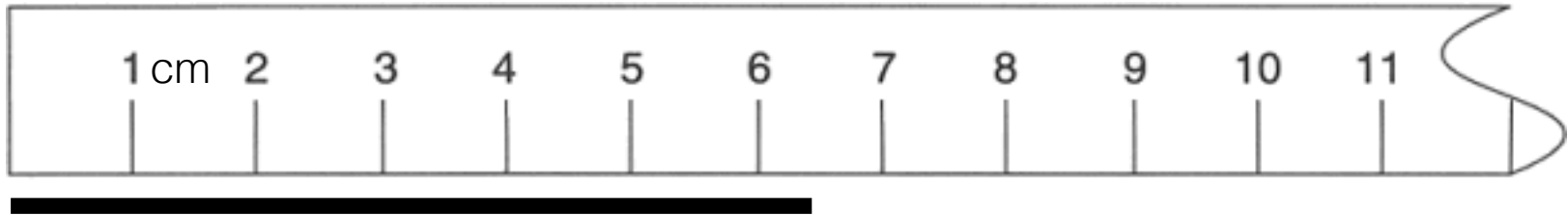
Not Accurate  
Not Precise



Precision: The closeness of a set of measurements to each other

Accuracy: The closeness of measurements to the correct or accepted value

Significant Figures (“sig figs”): The number of all known digits reported in a measurement, plus one estimated digit.



**What should be the reported length of the black line?**

*Draw this diagram!*

**To be completed in class!  
(leave space to write a measurement)**

# Rules for Determining the Number of Sig Figs in a Measurement

## All nonzero digits are significant

**721** mm

3 sig figs

**2,398** g

4 sig figs

**3.4** x 10<sup>-4</sup> s

2 sig figs

# Rules for Determining the Number of Sig Figs in a Measurement

**Zeros that are between nonzero digits are significant.**

**106** K

3 sig figs

**2,001** kg

4 sig figs

# Rules for Determining the Number of Sig Figs in a Measurement

**Zeros alone on the left are never significant (place holders).**

0.0**53** m

2 sig figs

0.0000**2** s

1 sig fig

# Rules for Determining the Number of Sig Figs in a Measurement

**Zeros alone on the right are significant if there is a decimal point somewhere in the number (accuracy).**

**121**0 m (place holder)                      3 sig figs

**200.** K (accuracy)                              3 sig figs

**80.00** g (accuracy)                              4 sig figs

**Counting numbers and exact numbers  
have an infinite number of significant  
figures**

10 pairs

$\infty$  sig figs

76 dogs

$\infty$  sig figs

1 ft = 12 in

$\infty$  sig figs

(conversion factor)



# Sig Fig Practice

How many sig figs are in each of the following measurements?

a. 28.6 g

b. 2440. cm

c. 910 m

d. 0.04604 L

e. 0.0067000 kg

**To be completed in class!  
(leave space to the right of each  
number)**

# Scientific Notation

only *one* digit to  
the left of the  
decimal

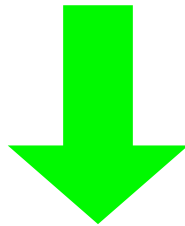
$$N \times 10^n$$

a power  
of 10

# Scientific Notation

Ex: Express 1234.56 in scientific notation.

1234.56



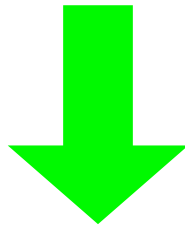

$1.23456 \times 10^3$

**if positive  
exponent,  
number is  
bigger than 1**

# Scientific Notation

Ex: Express 0.00657 in scientific notation.

0.00**6**57



6.57 x 10<sup>-3</sup>

**if negative  
exponent,  
number is  
smaller than 1**

# Scientific Notation Practice

Express the following measurements in scientific notation. Be sure to maintain the number of significant figures.

a. 5,453,000 km

b. 300.8 g

c. 0.00536 mm

d. 0.0120325 mg

e. 34,800 s

**To be completed in class!**  
**(leave space to the right of each number)**