

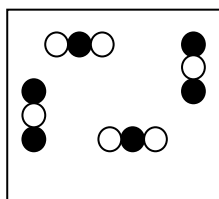
Complete the following assignment in your class notebook with the heading:

## Classifying Matter

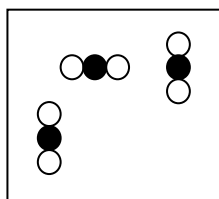
Classify each of the following as elements (E), compounds (C), homogeneous mixtures (Hom), or heterogeneous mixtures (Het). Write the letter X if it is none of these.

<u>E</u> Diamond (C)	<u>C</u> Sugar (C <sub>6</sub> H <sub>12</sub> O <sub>6</sub> )	<u>Hom</u> Milk	<u>E</u> Iron (Fe)
<u>Hom</u> Air	<u>C</u> Sulfuric Acid (H <sub>2</sub> SO <sub>4</sub> )	<u>Hom</u> Gasoline	<u>X</u> Electricity
<u>E</u> Krypton (Kr)	<u>E</u> Bismuth (Bi)	<u>E</u> Uranium (U)	<u>Het</u> Popcorn
<u>C</u> Water (H <sub>2</sub> O)	<u>C</u> Alcohol (CH <sub>3</sub> OH)	<u>Het</u> Pail of Garbage	<u>Het</u> A dog
___ Ammonia (NH <sub>3</sub> )	___ Salt (NaCl)	___ Energy	___ Gold (Au)
___ Wood	___ Bronze	___ Ink	___ Pizza
___ Dry Ice (CO <sub>2</sub> )	___ Baking Soda (NaHCO <sub>3</sub> )	___ Titanium (Ti)	___ Concrete

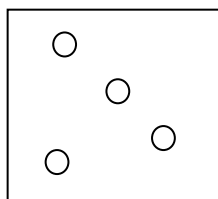
Match each diagram with its correct description. Diagrams will be used once.



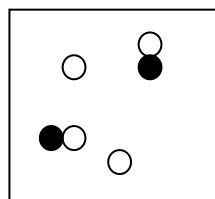
A



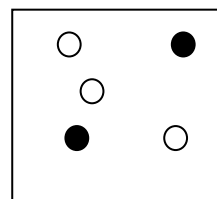
B



C



D



E

- C 1. Pure Element - only one type of atom present.
- E 2. Mixture of two elements - two types of uncombined atoms present.
- \_\_\_ 3. Pure compound - only one type of compound present.
- \_\_\_ 4. Mixture of two compounds - two types of compounds present.
- \_\_\_ 5. Mixture of a compound and an element.