

Complete the following assignment in your class notebook with the heading: Naming Acids

	Anion Formula	Anion Name	Acid Formula	Acid Name
Ex	$\text{ClO}^-$	hypochlorite	$\text{HClO}$	hypochlorous acid
1	$\text{Br}^-$			
2	$\text{SO}_3^{2-}$			
3			$\text{H}_2\text{CO}_3$	
4		iodide		
5	$\text{CrO}_4^{2-}$			
6			$\text{HClO}_4$	
7	$\text{NO}_3^-$			
8				hydrosulfuric acid
9			$\text{HF}$	
10				oxalic acid
11		arsenate		
12				sulfuric acid
13				acetic acid
14	$\text{Cl}^-$			
15		chlorate		
16			$\text{HCN}$	