Name:

## WP Practice Exam 1: Measurement

(Also review Math Essentials packet)

- 1. Physical chemical changes
  - a. An egg is cooked on the stove. (circle one)

Physical change Chemical change

b. A piece of paper is ripped into small pieces. (circle one)

Physical change Chemical change

- c. The frame of a bicycle begins to rust. (circle one)
  - Physical change Chemical change
- 2. Counting significant figures (see Math Essentials packet)
- 3. Rounding significant figures (see Math Essentials packet)
- 4. Scientific notation (see Math Essentials packet)
- 5. Precision vs accuracy
  - a. Determine whether each of the following results are accurate and/or precise:



- 6. Metric conversions (see Math Essentials packet)
- 7. Temperature conversions
  - a. Convert 23°C to Kelvin.
  - b. Convert 83°C to Fahrenheit.





- 8. Calculations with significant figures (see Math Essentials packet)
- 9. Word problems
  - a. You pull into a gas station in Switzerland and fill your rental car with gas. It takes 60.4 L of gasoline and they charge you 105 Swiss Francs. How much were you paying in dollars per gallon? Current exchange rate is 1.38 SF/dollar. (Tutorial vid available!)
  - b. Suppose that a standard snail's pace is measured to be 0.040 ft per min. Measured in cm per second, what is the value of the snail's pace?
  - c. Calculate the density, in g/cm<sup>3</sup>, of a 34.8 g rectangular solid object that measures 3.4 cm by 3.9 cm by 2.9 cm. Will the object sink or float in water? (Tutorial vid available!)
  - d. Calculate the volume, in mL, of 34 g of carbon tetrachloride (d = 1.59 g/mL).
  - e. Calculate the volume, in quarts, of 439 mg of ethyl alcohol. Density = 0.789 g/mL.
  - f. A piece of aluminum foil ( $d_{AI} = 2.70 \text{ g/cm}^3$ ) measures 10.4 cm by 12.6 cm and has a mass of 1.334 g. Calculate the thickness, in mm, of the aluminum foil. (Tutorial vid available!)