## Warm-Up

For each pair of substances listed below:

- i. Draw two Lewis structures for each and a dotted line to represent the intermolecular force of attraction between them. (See the examples worked out in class!)
- ii. Label the intermolecular force.
- iii. Circle the one that would have the higher boiling point.

a. HCN CH₃OH

b.  $F_2$   $Cl_2$ 

c.  $NF_3$   $C_2H_6$